Preparation of 7 β **-Methyldihydrothebaine**- ϕ^1

David L. Leland, Joseph O. Polazzi, and Michael P. Kotick*

Chemistry Department, Corporate Research Divison, Miles Laboratories, Inc., Elkhart, Indiana 46515

Received April 28, 1980

The reaction of thebaine (1) with lithium dimethylcuprate yields 7β -methyldihydrothebaine- ϕ (2). Acid hydrolysis of 2 gave mixtures of 7,8-didehydro-7,17-dimethyl-4-hydroxy-3-methoxy-B/C-cis- (4a) and -B/C-trans- (4b) morphinan-6-ones which, after separation, were hydrogenated to the saturated ketones 5a,b. The B/C ring junctures in 5a,b, respectively, were determined by mass spectral and chemical studies. Reaction of dihydrothebaine (6) or dihydrocodeinone enol acetate (8) with Me₂CuLi gave 4,5-epoxy cleaved 7 β -methylmorphinane derivatives, some of which could be converted to 5a.

We have recently reported that reaction of codeinone with lithium dialkylcuprates yields mainly 8β -alkylateddihydrocodeinones.² A 4,5-epoxy cleaved product, thebainone A,³ was found as a minor constituent of these reaction mixtures, particularly in the reaction of codeinone with lithium dimethylcuprate. With the goal of preparing compounds for further synthetic work in which the 4,5epoxy bond has been cleaved,^{4,5} we investigated the reaction of lithium dimethylcuprate with thebaine (1). We have unexpectedly found that this reaction yields 7β methyldihydrothebaine- ϕ (2).

Addition of a benzene solution of 1 to 1.2 equiv of ethereal lithium dimethylcuprate gave 2 as the major product (>90% by TLC), which was obtained in crystalline form as the monochloroform solvate. Chromatography of the mother liquors allowed isolation of the minor product, the 7α -methyl isomer 3, in about 4% yield. The NMR signal for the 7β -methyl group of 2 was observed as a doublet, centered at δ 1.17, whereas the corresponding signal for 3 was observed at δ 1.08. This difference in position for the 7-methyl signal is due to the anisotropic effect of the aromatic A ring and reflects our earlier observations.^{1,5} This anisotropic effect is also responsible for the low-field position (δ 6.13) observed for H-5.⁶

Treatment of 2 with aqueous acetic acid at 90–100 °C gave an approximately 3:1 mixture of the B/C-cis (4a) and B/C-trans (4b) α,β -unsaturated ketones (Scheme I) whereas hydrolysis with 5% hydrochloric acid gave a 94% yield of 4b. Conditions could not be found which gave exclusively the cis isomer 4a which corresponds to the natural morphine B/C configuration. Catalytic reduction of these separated isomers gave the 4-hydroxy-6-ones 5a and 5b. The assignment of B/C ring junctures in 5, and thus in 4, was initially based on the characteristic m/e 59 ion which is found only in the mass spectrum of ring C saturated cis-morphinanes and the relative abundance of molecular ions (trans > cis).⁷ Final confirmation of this assignment was demonstrated by conversion of 8 to 5a as described below.

It was reported some years ago that both dihydrothebaine $(6)^8$ and dihydrocodeinone enol acetate $(8)^9$ react with methyl Grignard reagents. These reactions yield, after workup including acid hydrolysis, mainly 5-methyldihydrothebainone (5,17-dimethyl-4-hydroxy-3-methoxymorphinan-6-one) together with a small percentage of the corresponding 7-methyldihydrothebainone.¹⁰ In view of this result, we further examined the reaction of Me₂CuLi with these substrates. Only methylation at the 7 position was observed in our reactions. Reaction of 6 with Me₂CuLi gave the 4-hydroxy-5,6-

didehydro-6-methoxy- 7β -methyl compound 7 in about 77% yield. Mild acid hydrolysis of 7 gave the saturated ketone 5a, identical with material prepared from 4a. A mixture of four products was obtained upon reaction of 8 with Me₂CuLi under similar conditions followed by mild basic workup. Two isomeric 6-oxo compounds were obtained in about equal amounts. One of these was 5a (NMR $\delta 0.87$ (7 α -methyl)) while the other was presumed to be the less stable (axial) 7β methyl compound 9. NMR confirmation of this supposition was obtained when a CDCl₃ solution of 9 upon treatment with CF₃COOH converted to a spectrum identical with that of 5a. 5.6-Didehydro-7-methyl compound 10 and the product of epoxy bond cleavage without alkylation 11 were also obtained in this reaction. Thus, Me₂CuLi reacts with 6 and 8 in a manner opposite to that of methyl Grignard.

The mechanism of reaction of lithium dialkylcuprates with various substrates remains a subject of vigorous investigation.¹¹ Our observed reaction of Me₀CuLi with enol ethers derived from the morphine alkaloids may proceed either by electron transfer or nucleophillic addition. Alkylation of the intermediates involved in these reactions takes place stereospecifically from the less hindered β face of the molecule.

The novel method for the introduction of a C-7 methyl group into the morphinane nucleus has allowed us to extend our previous studies aimed at determining the effect of alkyl substitution in the C ring of opiate derivatives on analgesic-narcotic antagonist activity. These studies will be reported elsewhere. 12

Experimental Section

Methods have previously been described.² Processing in the usual fashion implies that the organic phases were washed with dilute NH_4OH , dried (MgSO₄), and evaporated at 40 °C. The residue was further dried at 50-60 °C under high vacuum. Column chromatography was carried out over silica gel G (E. Merck), using the indicated amount of gel and the indicated CHCl₃-MeOH mixtures containing 0.5-1% (v/v) concentrated NH₄OH as eluant. NMR spectra were recorded in CDCl₃ unless otherwise stated.

⁽¹⁾ Analgesic Narcotic Antagonists. 3. A portion of this work was presented before the 180th National Meeting of the American Chemical Society, San Francisco, CA, Aug 1980; MEDI 59.

⁽²⁾ Kotick, M. P.; Leland, D. L.; Polazzi, J. O.; Schut, R. N. J. Med. Chem. 1980, 23, 166.
(3) Sawa, Y. K.; Horiuchi, M.; Tanaka, K. Tetrahedron 1965, 21, 1133.

⁽⁴⁾ Razdan, R. K.; Portlock, D. E.; Dalzell, H. C.; Malmberg, C. J. Org. Chem. 1978, 43, 3604.

⁽⁵⁾ Polazzi, J. O.; Schut, R. N.; Kotick, M. P.; Howes, J. F.; Osgood,
P. F.; Razdan, R. K.; Villarreal, J. E. J. Med. Chem. 1980, 23, 174.
(6) Stuart, K. L. Chem. Rev. 1971, 71, 47. This review discusses the related morphinandienone alkaloids. Many of these compounds are

<sup>reported as crystalline solvates.
(7) Mandelbaum, A.; Ginsberg, D. Tetrahedron Lett. 1965, 2479.
(8) Small, L.; Fitch, H. M.; Smith, W. E. J. Am. Chem. Soc. 1936, 58,</sup>

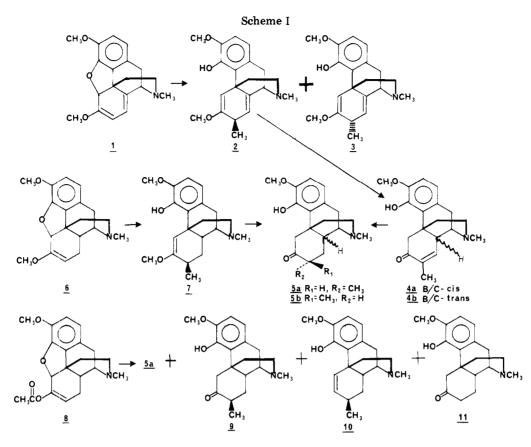
^{1457.}

⁽⁹⁾ Small, L.; Turnbull, S. G.; Fitch, H. M. J. Org. Chem. 1938, 3, 204.

⁽¹⁰⁾ The positon of the methyl group in the minor isomer was defi-nitively established by Stork and Bauer (J. Am. Chem. Soc. 1953, 75, 4373).

^{(11) (}a) House, H. O.; Snoble, K. A. J. Org. Chem. 1976, 41, 3076. (b) Casey, C. P.; Cesa, M. C. J. Am. Chem. Soc. 1979, 101, 4236. (12) Analgesic Narcotic Antagonists. Part 4. See: Leland, D. L.;

Kotick, M. P. J. Med. Chem., in press.



This method was also used to confirm the presence and amount of solvent of crystallization. Optical rotations were determined by using a Perkin-Elmer Model 241 polarimeter. Mass spectra were determined by using a Hewlitt-Packard 5985A GC/MS system and are reported as m/e (relative intensity). Only selected, significant peaks are reported.

3,6-Dimethoxy-78,17-dimethyl-4-hydroxy-5,6,8,14-tetradehydromorphinane (2). To a solution of Me₂CuLi, prepared from CuI (23.18 g, 125 mmol) and MeLi (250 mmol, 126 mL of a 1.8 M solution in Et₂O containing LiBr), in Et₂O (500 mL) stirred in an ice-salt bath under an argon atmosphere was added rapidly in a thin stream a solution of 1 (31.14 g, 100 mmol) in C_6H_6 (500 mL). The resulting suspension was stirred for 1 h in the cold, then poured into saturated NH4Cl solution (600 mL), and stirred for 15 min. The organic layer was separated and the aqueous phase adjusted to pH 13-14 by use of 50% NaOH. The aqueous phase was extracted with CHCl₃ and the organic phases were processed in the usual manner. Evaporation gave a foam which crystallized from CHCl₃ with the addition of hexane to give 33.20 g (74%) of 2 as the CHCl₃ solvate, mp 97-100 °C. Recrystallization from the same solvent pair gave pure 2. CHCl₃: mp 98-101.5 °C; NMR δ 7.30 (s, 1, CHCl₃), 6.65 (m, 2, H-1 and H-2), 6.13 (s, 1, H-5), 5.47 (d, 1, H-8, $J_{7,8} = 3$ Hz), 3.86 (s, 3-OCH₃), 3.63 (s, 6-CH₃), 2.33 (s, NCH₃), 1.17 (d, 3, 7 β -CH₃, $J_{7H,7CH_3} = 7$ Hz), ~6.20 (exchangeable 4-OH); mass spectrum, m/e 327 (M⁺, 54), 312 (100).

Anal. Calcd for $\hat{C}_{20}H_{25}N\hat{O}_{3}$ ·CHCl₃: C, 56.46; H, 5.87; N, 3.13. Found: C, 56.27; H, 5.82; N, 3.07.

3,6-Dimethoxy- 7α ,17-dimethyl-4-hydroxy-5,6,8,14-tetradehydromorphinane (3). The mother liquor obtained above was evaporated to a dry residue and chromatographed (500 g, 15:1:1%). After elution of additional 2 (2.46 g), 3 (1.36 g, 4%) was obtained as a foam; NMR δ 6.69 (m, 2, H-1 and H-2), 6.08 (br s, 1, H-5), 5.57 (d, 1, H-8, $J_{7,8} = 4$ Hz), 3.87 (s, 3-OCH₃), 3.65 (s, 6-OCH₃), 2.42 (s, NCH₃), 1.08 (d, 3, 7α -CH₃, $J_{7H,7CH_3} = 7$ Hz).

7,8-Didehydro-7,17-dimethyl-4-hydroxy-3-methoxy-B/Ccis- (4a) and -B/C-trans- (4b) morphinan-6-ones. A. To HOAc (150 mL), preheated in an oil bath to 90 °C, was added 2-CHCl₃ (10.00 g, 22.4 mmol) and the mixture kept between 90–100 °C for 40 min. The cooled solution was made basic by the addition of NH₄OH and extracted with several portions of CHCl₃. Processing in the usual fashion gave a residue (8.36 g) which was chromatographed (500 g, 20:1:0.5%). Eluted first from the column was 4b (1.61 g, 23%) obtained as a tan glass: NMR δ 6.65 (3, H-1, H-2, and 4-OH), 6.26 (br s, 1, H-8), 4.10 (d, 1, H-5 α , J = 17 Hz), 3.85 (OCH₃), 2.36 (NCH₃), 1.86 (m, 3, 7-CH₃); mass spectrum, m/e 313 (M⁺, 56), 176 (100).

Next eluted was 4a which was obtained as crystals (5.07 g, 72%) after evaporation. Recrystallization from EtOAc gave white needles of 4a, mp 183–185 °C, which were found to contain 0.33 mol of EtOAc by NMR and elemental analysis: NMR δ 6.58 (H-1 and H-2), 6.40 (br s, 2, H-8 and 4-OH), 4.30 (d, 1, H-5 α , J = 15 Hz), 3.76 (OCH₃), 2.40 (NCH₃), 1.60 (m, 3, 7-CH₃); mass spectrum, m/e 313 (M⁺, 74), 176 (100), 109 (59).

Anal. Calcd for $C_{19}H_{23}NO_3 \cdot 0.33C_4H_8O_2$: C, 71.25; H, 7.55; N, 4.09. Found: C, 71.21; H, 7.58; N, 3.98.

B. Compound 2 CHCl₃ was hydrolyzed at 90–100 °C as above, but with 5% aqueous HCl. The residual foam was chromatographed to yield, after evaporation of appropriate fractions, 6.60 g (94%) of 4b as a foam followed by 0.40 g (6%) of 4a. Two crystallizations of 4b from Et₂O gave 3.0 g of shiny tan crystals, mp 135–138 °C after drying at 50 °C in high vacuum, which were found to contain 0.33 mol of Et₂O by NMR and elemental analysis.

Anal. Calcd for $C_{19}H_{23}NO_3^{*}0.33C_4H_{10}O$: C, 72.23; H, 7.85; N, 4.14. Found: C, 72.30; H, 7.52; N, 4.33.

 7α ,17-Dimethyl-4-hydroxy-3-methoxy-B/C-cis-morphinan-6-one (5a). A solution of 4a-0.5C₄H₈O₂ (22.0 g, 61 mmol) in 95% EtOH (200 mL), containing concentrated HCl (12 mL) and 10% Pd/C (2.0 g), was hydrogenerated at an initial pressure of 50 psi until uptake of H_2 ceased. The mixture was filtered from the catalyst and the filtrate evaporated to a small volume. The residue was diluted with H₂O, made basic with NH₄OH, and extracted with CHCl₃. The CHCl₃ extracts were processed to give a thick syrup which crystallized on addition of EtOAc. The crystals were collected and air-dried to give 14.4 g (74%) of 5a as white needles, mp 163-165 °C: NMR δ 6.55 (H-1 and H-2), 6.40 (br, 4-OH), 4.22 (d, H-5 α , J = 13 Hz), 3.77 (OCH₃), 2.38 (NCH₃), 0.87 (d, 3, 7α -CH₃, J = 6.5 Hz); mass spectrum, m/e 315 (M⁺, 32), 178 (100), 150 (15), 115 (16), 59 (12). Recrystallization of 5a from acetone gave the hemiacetone solvate of 5a, mp 166-167 °C (lit.⁸ mp 168–168.5 °C), $[\alpha]_D = 55^\circ$ (c 1.0, CHCl₃) (lit.⁸ $[\alpha]_D = 57^\circ$ [c 1.0, EtOH]). The oxime of 5a melted at 190-191 °C (lit.⁸) 191-192 °C).

Anal. Calcd for $C_{19}H_{25}NO_3$.0.5CH₃COCH₃: C, 71.48; H, 8.19; N, 4.07. Found: C, 71.69; H, 8.27; N, 4.12.

7α,17-Dimethyl-4-hydroxy-3-methoxy-B/C-trans-morphinan-6-one (5b). Compound 4b was hydrogenated in a similar manner to that reported for 4a. Workup gave a foam which was chromatographed to give 5b as a foam: NMR δ 6.68 (H-1 and H-2), 6.20 (br, 4-OH), 4.09 (s, 0.5, 1/2 doublet for H-5 α), 3.87 (unsymmetrical s, 3.5, 1/2 doublet for H-5 α , OCH₃), 2.32 (NCH₃), 1.02 (unsymmetrical d, 3, 7β -CH₃, J = 5 Hz); mass spectrum, m/e315 (M⁺, 80), 272 (55), 244 (100), 178 (76), 122 (53), 115 (31), 59 (0). Conversion to the HCl salt followed by two crystallizations from MeOH-EtOAc gave an analytical sample of 5b·HCl, mp sinters 142-145 °C, melts 222-224 °C.

Anal. Calcd for C₁₉H₂₅NO₃·HCl: C, 64.86; H, 7.45; N, 3.98. Found: C, 64.47; H, 7.73; N, 4.21.

5,6-Didehydro-3,6-dimethoxy-7 β ,17-dimethyl-4-hydroxymorphinane (7). A solution of Me₂CuLi (12.5 mmol) was prepared in Et₂O (75 mL) as previously described. To this was added rapidly dropwise a solution of dihydrothebaine (6, 3.13 g, 10 mmol) in benzene (75 mL) and the reaction mixture was stirred for 1 h in the ice bath. The mixture was poured into saturated NH₄Cl solution, stirred for 30 min and then adjusted to pH 11 with NH₄OH. Extraction with CHCl₃ followed by processing in the usual fashion gave 3.76 g of a foam which was chromatographed (400 g, 10:1:1%). Fractions containing the major product, homogeneous by TLC, were combined and evaporated to give 7 (2.54 g, 77%) as a foam; NMR & 6.60 (s, H-1 and H-2), 6.13 (br, 4-OH), 5.68 (s, 1, H-5), 3.83 (3-OCH₃), 3.58 (6-OCH₃), 2.40 (NCH₃), 1.18 $(m, 3, 7\beta - CH_3).$

Hydrolysis of 7 to 5a. Compound 7 (2.54 g, 7.7 mmol) in 1 N HCl (30 mL) was heated on the steam bath for 30 min. The cooled solution was made basic with NH₄OH and extracted with CHCl₃. Processing followed by evaporation gave a crystalline residue. Recrystallization from acetone gave 5a.0.5C₃H₆O (1.29 g, 48%), mp 166-167 °C, identical with material prepared above.

Reaction of Dihydrocodeinone Enol Acetate (7) with Me_2CuLi . A solution of 8 (6.40 g, 18.75 mmol) in benzene (100 mL) was added rapidly dropwise to a solution of Me₂CuLi (40 mmol), prepared in Et₂O (200 mL), at 0 °C under argon. The mixture was stirred for 1 h at 0 °C and poured into saturated NH₄Cl solution and the pH was adjusted to 10 with NH₄OH. After the mixture was stirred for 30 min, processing of the combined organic phase and CHCl₃ extracts gave a foam which contained four major compounds as indicated by TLC. The foam was chromatographed (750 g, 15:1:0.75%) and fractions were pooled on the basis of TLC.

First eluted was 5,6-didehydro-7 β ,17-dimethyl-4-hydroxy-3methoxymorphinane (10, 1.48 g, 26%): NMR δ 6.63 (2.5, H-1, H-2, and 1/2 doublet for H-5), 6.46 (1/2 unsymmetrical doublet for H-5), 6.00 (4-OH), 5.55 (pair of d, 1, H-6, $J_{5,6} = 10$ Hz, $J_{6,7}$ = 3.5 Hz), 3.83 (OCH₃), 2.45 (NCH₃), 1.05 (unsymmetrical d, 7-CH₃, J = 7 Hz). This material was converted to the HCl salt which was crystallized from EtOAc to give 10·HCl, mp >250 °C dec.

Anal. Calcd for C₁₉H₂₅NO₂·HCl: C, 67.94; H, 7.80; N, 4.17. Found: C, 67.70; H, 7.89; N, 4.07.

Next eluted was 7β , 17-dimethyl-4-hydroxy-3-methoxymorphinan-6-one (9, 1.71 g, 28%); NMR δ 6.62 (H-1 and H-2), 4.10 (H-5 α , J = 15 Hz), 3.82, 2.43, 1.21 (d, 7 β -CH₃, J = 7 Hz). Two recrystallizations from EtOAc gave the hemi-EtOAc solvate of 9, mp 163-165 °C, $[\alpha]_D$ -115° (c 1.0, CHCl₃).

Anal. Calcd for $C_{19}N_{25}NO_3 \cdot 0.5C_4H_8O_2$: C, 70.17; H, 8.13; N, 3.90. Found: C, 70.44; H, 8.19; N, 3.87.

Continued elution gave 5a (1.66 g, 27%) followed by 11 (0.65 g, 12%) which was identified by comparison with an authentic sample.

Acknowledgment. We are indebted to Drs. R. N. Schut and J. E. Villarreal for their continued interest and encouragement during the course of this work.

Registry No. 1, 115-37-7; 2, 74466-73-2; 3, 74466-74-3; 4a, 74466-75-4; 4b, 74497-87-3; 5a, 74497-88-4; 5b, 74497-89-5; 5b·HCl, 74497-90-8; 6, 57281-79-5; 7, 74466-76-5; 8, 466-90-0; 9, 74497-91-9; 10, 74466-77-6; 10·HCl, 74466-78-7; 11, 847-86-9.

Two New Germacranolides from *Melampodium leucanthum* and Their **Reductive and Oxidative Rearrangements**

Errol J. Olivier, Donald L. Perry, and Nikolaus H. Fischer*

Department of Chemistry, Louisiana State University, Baton Rouge, Louisiana 70803

Received November 14, 1979

The isolation and structure elucidation of two new 4,5-dihydrogermacranolides, 9-acetoxymelnerin A and B, from Melampodium leucanthum Torr. and Gray (Compositae, Heliantheae) are reported. Reduction of the two germacranolides with sodium borohydride proceeds, besides saturation of the lactonic exocyclic methylene function, under 1(10) to 9,10 double bond rearrangment with the loss of the C-9 acetoxy group, a process which can be interpreted as a S_N' reaction. Pyridinium chlorochromate oxidation of the C-15 alcohol function of 9-acetoxymelnerin A provides the aldehyde as a minor product, the major product being a ketone formed by a C-4 to C-15 shift of C-5 of the ten-membered ring, resulting in an 11-membered-ring skeleton. The separation of the previously inseparable melnerin A and B mixture by reverse-phase high-pressure liquid chromatography is described, and the physical parameters of the pure compounds are reported.

In our biochemical systematic study of the genus Melampodium (Compositae, Heliantheae) we have in the past reported results of our populational analysis for sesquiterpene lactones in M. leucanthum. Our previous investigations of this chemically diverse species have led to the isolation of melampolides,¹⁻⁴ germacranolide dilactones,^{5,6} and *cis,cis*-germacranolides.⁶ Now we wish to describe the isolation, structure elucidation, and chemistry of two new 4,5-dihydrogermacranolides which are structurally related to melnerins A and B, compounds which have previously

 ^{(1) (}a) Fischer, N. H.; Wiley, R.; Wander, J. D. J. Chem. Soc., Chem. Commun. 1972, 137. (b) Neidle, S.; Rogers, D. Ibid. 1972, 140.
 (2) Fischer, N. H.; Wiley, R. A.; Lin, H. N.; Karimian, K.; Politz, S. M. Phytochemistry 1975, 14, 2241.

⁽³⁾ Fischer, N. H.; Wiley, R. A.; Perry, D. L.; Haegele, K. D. J. Org. Chem. 1976, 41, 3956.

⁽A) Watkins, S. F.; Korp, J. D.; Bernal, I.; Perry, D. L.; Bhacca, N. S.;
(Fischer, N. H. J. Chem. Soc., Perkin Trans. 2 1978, 599.
(5) Perry, D. L.; Fischer, N. H. J. Org. Chem. 1975, 40, 3480.
(6) Fischer, N. H.; Seaman, F. C.; Wiley, R. A.; Haegele, K. D. J. Org. Chem. 1978, 43, 4984.